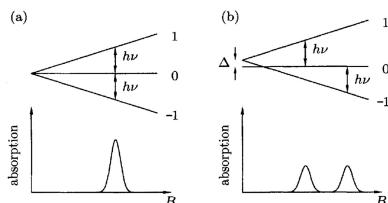


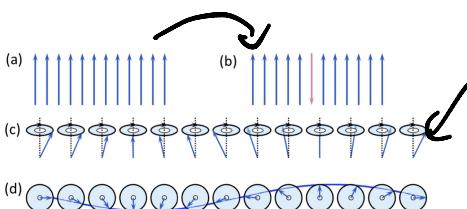
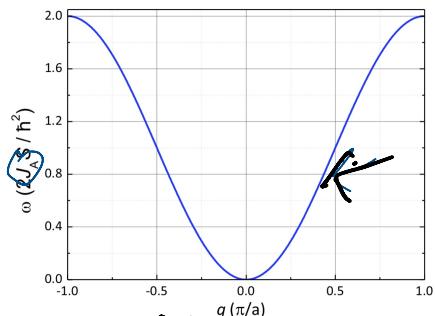
Lecture 28

Review

ESR:



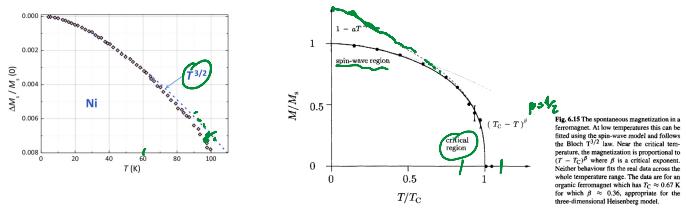
Spin waves:



Block $T^{3/2}$ law ↗

$$\frac{\Delta M}{M_s(T=0)} \propto \left(\frac{k_B T}{\lambda_A S / t_0} \right)^{3/2} \quad \text{Block } T^{3/2} \text{ law}$$

- Describes the LT-dependence of a ferromagnet well



Elastic neutron scattering

- neutron is a spin-1/2 particle, no charge
- non-zero magnetic moment.
- produced in great quantities by fission reactions inside the fuel elements of nuclear reactors.
- The beam of neutrons which emerges from a reactor has a spectrum of energies determined by the temperature T of the moderator

↳ slows down your neutrons (H_2O)

The maximum of this distribution is at

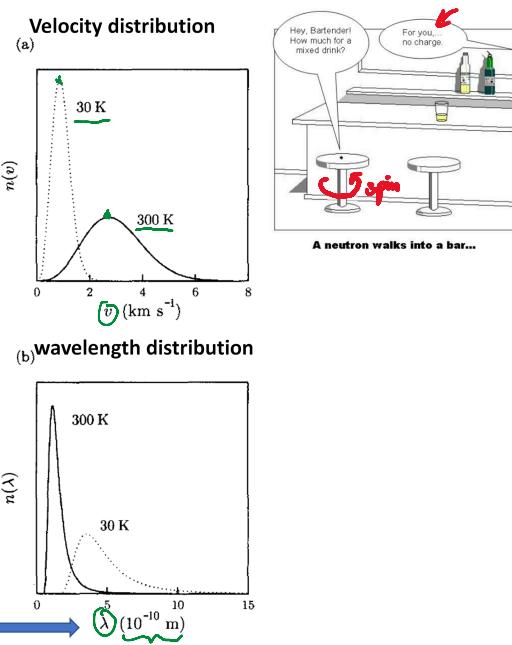
$$v = \sqrt{\frac{3k_B T}{m_n}}$$

$\propto v^3 \exp\left(-\frac{1}{2} \frac{m_n v^2}{k_B T}\right)$, Maxwell $\propto v^2 \cdot \exp\left(-\frac{1}{2} \frac{m_n v^2}{k_B T}\right)$
emission through a pin hole

The de Broglie wavelength λ of a neutron with velocity v is

$$\lambda = \frac{h}{m_n v}$$

On the order of atomic lattices



Elastic neutron scattering

- Neutrons scatter:
1. with nucleus via the strong nuclear force
 2. NOT with electron clouds (no charge)
 3. by variations in magnetic field within a crystal via the electromagnetic interaction
- Neutron couples to the magnetic moment of the electrons

1. nucleus via the strong nuclear force

- incident neutron wave produces a spherically symmetric elastically scattered wave
- elastically scattered neutrons can produce strong Bragg reflections

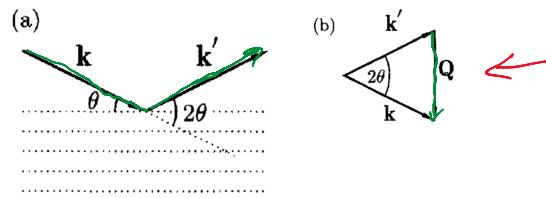
Setups:

- reactor (e)

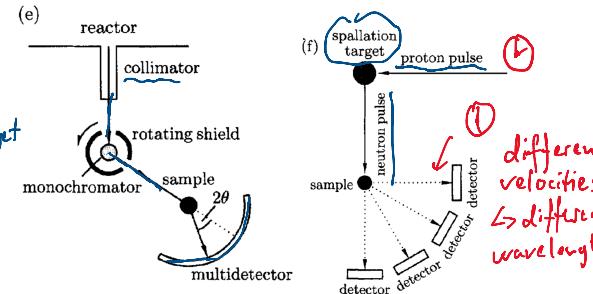
spallation source pulse of neutrons is produced by spallation (f) from a target. $\text{proton} \rightarrow \text{spallation target}$
Aufsplittung Determine neutron energy by time-of-flight measurements

2. Contrast to X-ray and electron scattering:

- Interaction with the nucleus, not the electron cloud
- Element-sensitive, isotope-sensitive
- highly penetrating probe
- does not damage bio-samples



Bragg reflections when scattering vector \vec{Q}
= reciprocal lattice vector \vec{g}



① different velocities
 ↳ different wavelength

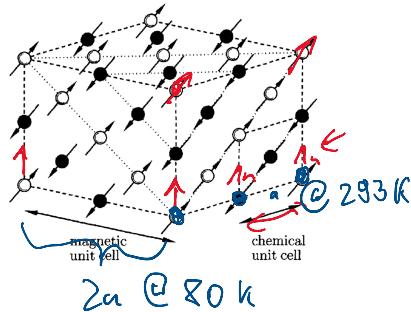
Elastic neutron scattering

3. But Neutron is also magnetic: magnetic scattering.

→ spin-up and spin-down electronic moments 'look' different

Example: magnetic structure of the antiferromagnet MnO.
 $T_N = 116 \text{ K}$

Amplitude of the magnetic Bragg peaks can be used as a measure of the strength of the magnetic order, and hence the magnetic order can be followed as a function of temperature.



$2\alpha @ 80 \text{ K}$

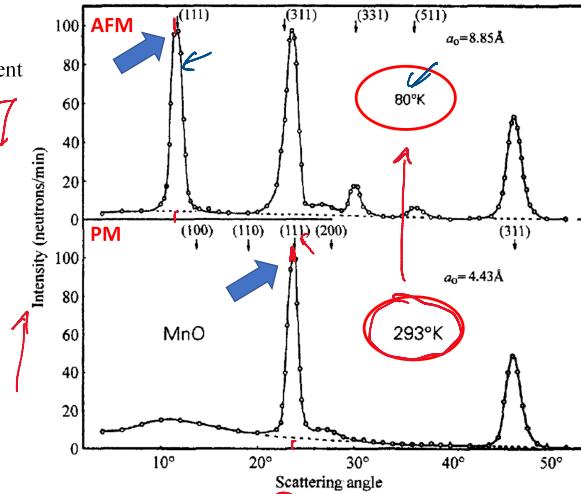


Fig. 5.19 Neutron diffraction patterns for MnO below and above T_N . After C. G. Shull, W. A. Strauser and E. O. Wollan, Phys. Rev., 83, 333 (1951).

Inelastic neutron scattering

The magnitude of the incident neutron wave vector k is now no longer equal to the magnitude of the scattered neutron wave vector k' .

$$E = \frac{\hbar^2 k^2}{2m_n} \quad E' = \frac{\hbar^2 k'^2}{2m_n}$$

Conservation of energy and momentum implies that

$$\begin{aligned} E &= E' + \hbar\omega && \text{phonon [boring]} \\ k &= k' + q + G, && \text{spin waves (not boring)} \end{aligned}$$

- measurement of k , k' , E and E' allows a determination of ω and q .

- Neutrons have energies similar to the energies of atomic and electronic processes, i.e. in the meV to eV range

- magnon energies are typically in the range 1-10 meV

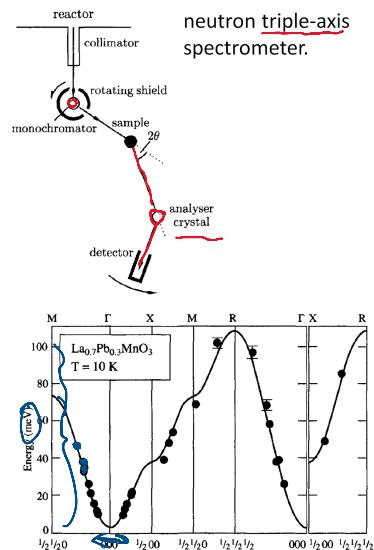
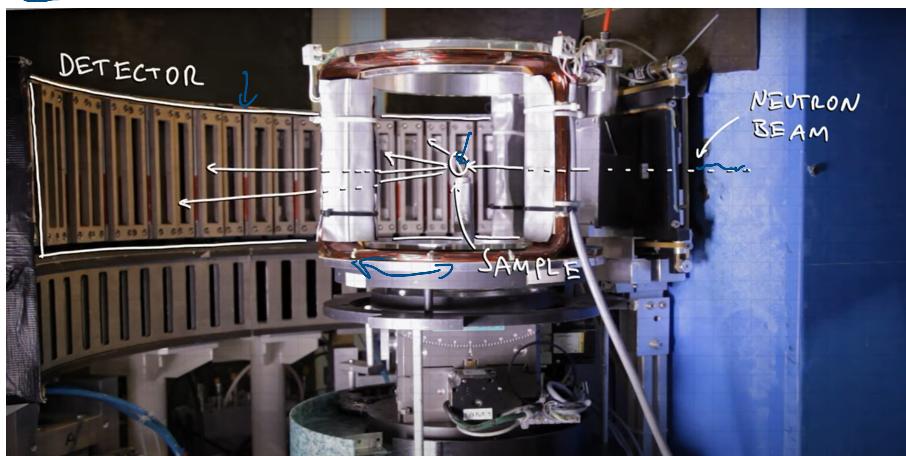
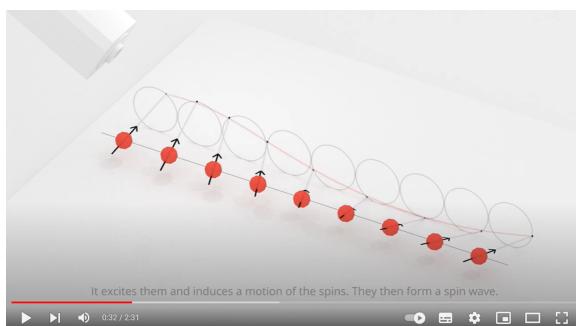


Fig. 6.18 Spin wave dispersion in the ferromagnetic oxide $\text{La}_{0.7}\text{Pb}_{0.3}\text{MnO}_3$ measured at 10 K by inelastic neutron scattering. The horizontal axis corresponds to the magnon wave vector. After Perring et al. 1996.

Jülich Centre for Neutron Science



Quantum Mechanics and Neutron Scattering 2 / 2



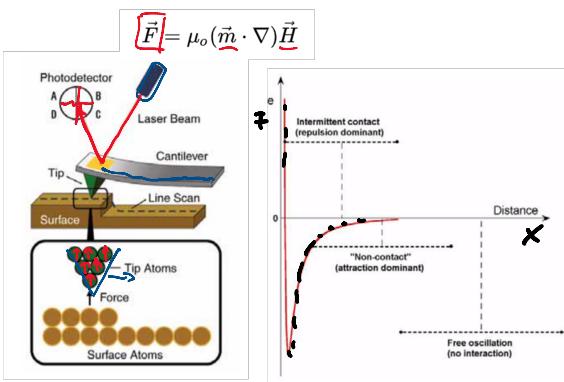
Neutron Inelastic Scattering

CNRS

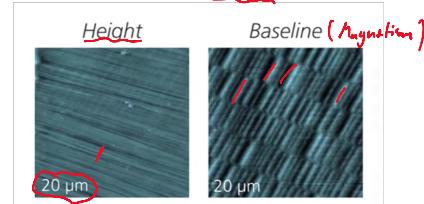
Scanning Probe Methods

Magnetic Force Microscopy

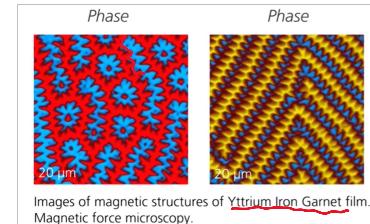
- Atomic Force Microscope: Probes the topography (height) of a surface using the force interaction with a probe tip
- Using a magnetic tip, magnetic signals can also be detected, due to an additional magnetic force



Topography and magnetic structure of a hard disc



Domain structure
Yttrium Iron Garnet (YIG) film



Scanning Probe Methods

Spin-polarized scanning tunneling microscopy

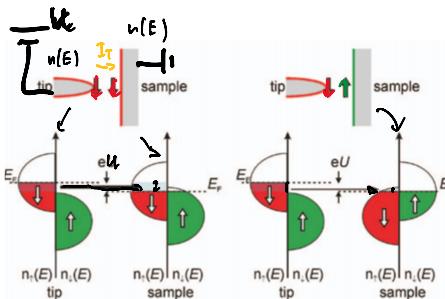
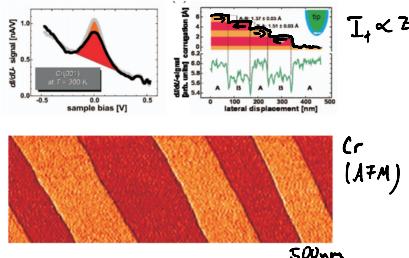
Spin-dependent conductance of a ferromagnet-tunnelbarrier-ferromagnet interface

$$G_{\uparrow\downarrow} = G_{fbf'}(1 + P_{fb}P_{f'b}),$$

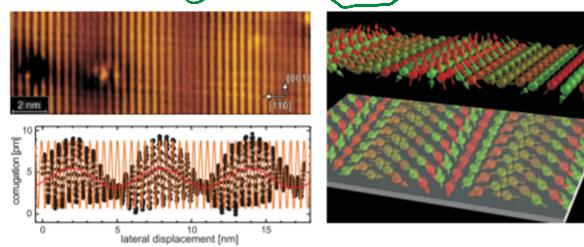
$$G_{\uparrow\downarrow} = G_{fbf'}(1 - P_{fb}P_{f'b}).$$

$$\frac{G_{\uparrow\downarrow} - G_{\uparrow\downarrow}}{G_{\uparrow\downarrow} + G_{\uparrow\downarrow}} = P_{fb}P_{f'b} =: P_{fbf'}$$

Fe-coated W tip above differently magnetized terraces of an antiferromagnetic Cr001 surface



Spin spiral state: Mn grown on a W(110) substrate DM



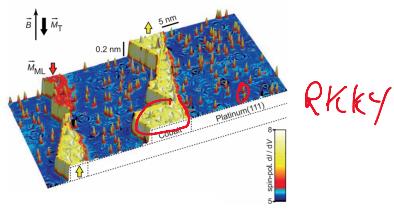
Roland Wiesendanger: Spin mapping at the nanoscale and atomic scale, Rev. Mod. Phys. 81 (2009)

Scanning Probe Methods

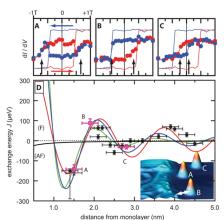
Spin-polarized scanning tunneling microscopy

Revealing Magnetic Interactions from
Single-Atom Magnetization Curves

Focke Meier,* Lihui Zhou, Jens Wiebe,† Roland Wiesendanger



SCIENCE VOL 320 4 APRIL 2008



Mößbauer Spectroscopy

Rudolf Mößbauer 1957/1958

(Nobel prize
1961)

- Also magnetic resonance technique, but a little bit different

- Recoil-free absorption of gamma rays emitted by one nucleus that is absorbed by a sample containing the same isotope

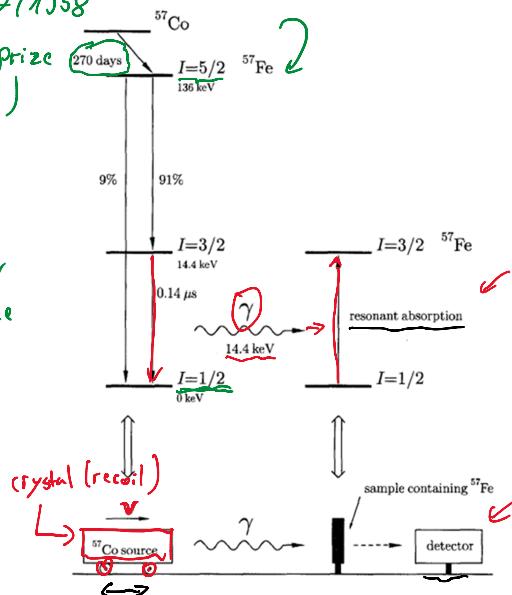
- Ingredients: a source of ^{57}Co → $^{57}\text{Fe}(I=5/2)$
→ decays to excited $^{57}\text{Fe}(I=3/2)$
→ decays to $^{57}\text{Fe}(I=1/2)$

- Emits a 14.4 keV gamma ray ($f = 3.5 \times 10^{18} \text{ Hz}$)

- To sweep frequency the Doppler effect is used ($v = 1 \text{ mm/s}$ corresponds to 12 MHz)

- Use ^{57}Fe in the sample as a resonant absorber

- Other elements are also possible, but not all ($^{113}\text{Sn}, ^{123}\text{I} \dots$)

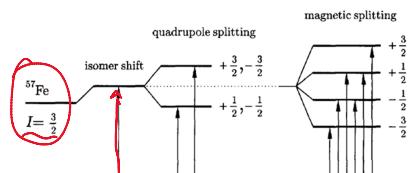


H		Element, bei dem der Mößbauer- effekt beobachtet wurde												He
Li	Be													
Na	Mg													
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi
Fr	Ra	Ac				Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy
						Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf

Mößbauer Spectroscopy

What can we measure with this?

1. Isomer shift (Comparable to Fermi contact basically)
→ shift of the resonant absorption due to differences



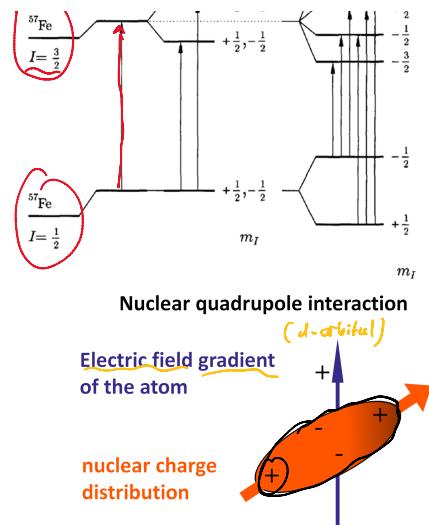
What can we measure with this?

1. Isomer shift (Comparable to Fermi contact basically)
 - shift of the resonant absorption due to differences in charge and Coulomb interaction
 - (Oxidation state, valency, electron shielding)
2. Nuclear Quadrupole splitting

$$H_{NQI} = \rho \cdot I_z^2$$

Coupling of the nuclear quadrupole moment to the charge gradient (e.g. d-orbitals)
(Oxidation State, Spin state, site symmetry and arrangement of ligands)

3. "Local" magnetic field
 - Exchange interaction
 - Hyperfine interaction



Mössbauer Spectroscopy

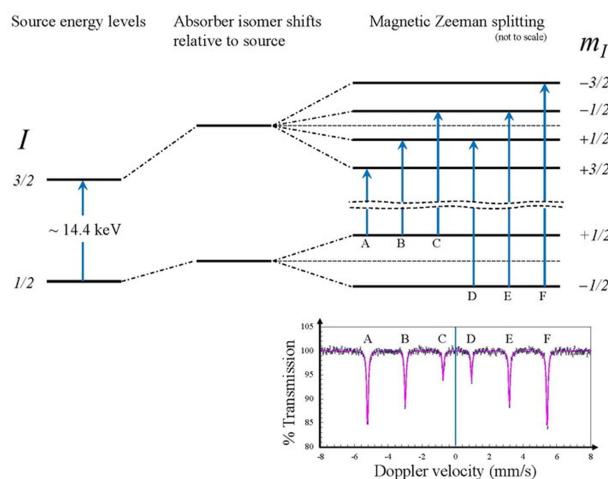


Fig. 4: Mossbauer spectrum and diagram illustrating magnetic Zeeman splitting in ^{57}Fe .

Mössbauer Spectroscopy

